CAMBRIDGE
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Cambridge International AS & A Level

PHYSICS **9702/43**

Paper 4 A Level Structured Questions **May/June 2025**

MARK SCHEME

Maximum Mark: 100

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the May/June 2025 series for most Cambridge IGCSE, Cambridge International A and AS Level components, and some Cambridge O Level components.

Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptions for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1: Marks must be awarded in line with: • the specific content of the mark scheme or the generic level descriptors for the question • the specific skills defined in the mark scheme or in the generic level descriptors for the question • the standard of response required by a candidate as exemplified by the standardisation scripts.
GENERIC MARKING PRINCIPLE 2: Marks awarded are always **whole marks** (not half marks, or other fractions).
GENERIC MARKING PRINCIPLE 3: Marks must be awarded **positively**: • marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate • marks are awarded when candidates clearly demonstrate what they know and can do • marks are not deducted for errors • marks are not deducted for omissions • answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.
GENERIC MARKING PRINCIPLE 4: Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5: Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).
GENERIC MARKING PRINCIPLE 6: Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1	Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
2	The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
3	Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
4	The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
5	<u>‘List rule’ guidance</u> For questions that require ***n*** responses (e.g. State **two** reasons …): • The response should be read as continuous prose, even when numbered answer spaces are provided. • Any response marked *ignore* in the mark scheme should not count towards ***n***. • Incorrect responses should not be awarded credit but will still count towards ***n***. • Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response. • Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

6	Calculation specific guidance Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'. For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values. For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme. Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.
7	Guidance for chemical equations Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme. State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Annotations guidance for centres

Examiners use a system of annotations as a shorthand for communicating their marking decisions to one another. Examiners are trained during the standardisation process on how and when to use annotations. The purpose of annotations is to inform the standardisation and monitoring processes and guide the supervising examiners when they are checking the work of examiners within their team. The meaning of annotations and how they are used is specific to each component and is understood by all examiners who mark the component.

We publish annotations in our mark schemes to help centres understand the annotations they may see on copies of scripts. Note that there may not be a direct correlation between the number of annotations on a script and the mark awarded. Similarly, the use of an annotation may not be an indication of the quality of the response.

The annotations listed below were available to examiners marking this component in this series.

Annotations

Annotation	Meaning
AE	arithmetic error
BOD	benefit of the doubt given
CON	contradiction in response, mark not awarded
✓	correct point or mark awarded
ECF	error carried forward applied
SF	error in number of significant figures
I	incorrect or insufficient point ignored while marking the rest of the response
XP	incorrect physics
✗	incorrect point or mark not awarded
^	information missing or insufficient for credit

Annotation	Meaning
M0	mandatory mark not awarded
SEEN	point has been noted, but no credit has been given or blank page seen
POT	power of ten error
TE	transcription error

Abbreviations

/	Alternative and acceptable answers for the same marking point.
( )	Bracketed content indicates words which do not need to be explicitly seen to gain credit but which indicate the **context** for an answer. The context does not need to be seen but if a context is given that is incorrect then the mark should not be awarded.
___	Underlined content must be present in answer to award the mark. This means either the exact word or another word that has the same technical meaning.

Mark categories

B marks	These are independent marks, which do not depend on other marks. For a **B** mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer.
M marks	These are mandatory marks upon which **A** marks later depend. For an **M** mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer. If a candidate is not awarded an **M** mark, then the later **A** mark cannot be awarded either.
C marks	These are compensatory marks which can be awarded even if the points to which they refer are not written down by the candidate, providing subsequent working gives evidence that they must have known them. For example, if an equation carries a **C** mark and the candidate does not write down the actual equation but does correct working which shows the candidate knew the equation, then the **C** mark is awarded. If a correct answer is given to a numerical question, all of the preceding **C** marks are awarded automatically. It is only necessary to consider each of the **C** marks in turn when the numerical answer is not correct.
A marks	These are answer marks. They may depend on an **M** mark or allow a **C** mark to be awarded by implication.

Question	Answer	Marks
1(a)	work done per unit mass	B1
	work (done in) moving mass from infinity (to the point)	B1
1(b)(i)	evidence of addition of 3.4×10^6 to 1.7×10^6 **or** 6.8×10^6	C1
	$GM \times 122 / (5.1 \times 10^6)$ **or** $GM \times 122 / (10.2 \times 10^6)$	C1
	$6.67 \times 10^{-11} \times M \times 122 \times [(5.1 \times 10^6)^{-1} - (10.2 \times 10^6)^{-1}] = 5.1 \times 10^8$ leading to $M = 6.4 \times 10^{23}$ kg	A1
1(b)(ii)	$\phi = (-)(6.67 \times 10^{-11} \times 6.4 \times 10^{23}) / (3.4 \times 10^6)$	C1
	$= -1.3 \times 10^7$ J kg^{-1}	A1
1(c)(i)	Mars takes (just under) 25 hours to rotate once on its axis	B1
1(c)(ii)	orbit is equatorial **or** orbit is in same direction as direction of rotation of Mars	B1

Question	Answer	Marks
2(a)	(electric) force is (directly) proportional to product of charges	B1
	force (between point charges) is inversely proportional to the square of their separation	B1
2(b)(i)	charge = (+)2*e*	A1
2(b)(ii)	$F = 2 \times (1.60 \times 10^{-19})^2 / [4\pi \times 8.85 \times 10^{-12} \times (170 \times 10^{-12})^2] = 1.6 \times 10^{-8}$ N	A1
2(c)(i)	$F = mv^2 / r$	C1
	$v = [(1.6 \times 10^{-8} \times 170 \times 10^{-12}) / (9.11 \times 10^{-31})]^{½}$ $= 1.7 \times 10^6$ m s^{-1}	A1
2(c)(ii)	$F = mr\omega^2$ **and** $\omega = 2\pi / T$ $F = 4\pi^2 mr / T^2$	C1
	$T = [(4\pi^2 \times 9.11 \times 10^{-31} \times 170 \times 10^{-12}) / (1.6 \times 10^{-8})]^{½}$ $= 6.2 \times 10^{-16}$ s	A1
	or	
	$v = 2\pi r / T$	**(C1)**
	$T = (2\pi \times 170 \times 10^{-12}) / (1.73 \times 10^6)$ $= 6.2 \times 10^{-16}$ s	**(A1)**

Question	Answer	Marks
2(d)(i)	$E \propto Q / r^2$	**C1**
	ratio = $[1.60 \times 10^{-19} \times (170 \times 10^{-12})^2] / [3.2 \times 10^{-19} \times (340 \times 10^{-12})^2]$ = 0.13	**A1**
2(d)(ii)	resultant force slightly less (than 1.6×10^{-8} N) so speed lower **or** resultant force slightly less (than 1.6×10^{-8} N) so period greater	**B1**

Question	Answer	Marks
3(a)	(thermal) energy per unit mass (to cause state change)	B1
	(thermal) energy to change state at constant temperature	B1
3(b)	(for vaporisation): involves greater change in volume (of substance) **or** involves greater increase in separation of molecules	B1
	more work has to be done by molecules (to separate) **or** greater increase in potential energy of molecules	M1
	kinetic energy of molecules unchanged, so more thermal energy needed	A1
3(c)	$Q = mc\Delta\theta$ **and** $Q = mL$	C1
	$\Delta\theta$ for the water = 26.4 – 10.3	C1
	$(37.0 \times L) + (37.0 \times 4.18 \times 10.3) = (208 \times 4.18 \times 16.1)$	C1
	$L = 335\ \text{J g}^{-1}$	A1

Question	Answer	Marks
4(a)(i)	sum of potential energy and kinetic energy	**B1**
	(total) energy of random motion of particles	**B1**
4(a)(ii)	potential energy (of molecules) (in an ideal gas) is zero, so the internal energy of the gas is equal to the total kinetic energy (of molecules)	**B1**
	kinetic energy of molecules is proportional to (thermodynamic) temperature (so internal energy is proportional to (thermodynamic) temperature))	**B1**
4(b)	cooling work done = 0	**B1**
	compression increase in internal energy = $+2U$	**B1**
	cooling change in internal energy = $-U$	**B1**
	both rows: thermal energy adds to work to give increase in internal energy in terms of U and/or W (if fully correct, thermal energy for compression = $2U - W$ and thermal energy for cooling = $-U$: <table><tr><td>compression</td><td>$+W$</td><td>$2U - W$</td><td>$+2U$</td></tr><tr><td>cooling</td><td>0</td><td>$-U$</td><td>$-U$</td></tr></table>)	**B1**

Question	Answer	Marks
5(a)(i)	amplitude = 0.60 m	**A1**
5(a)(ii)	oscillations are simple harmonic	**B1**
5(b)	*Any three points from:* • mean / equilibrium position is at h = 1.4 m • total energy of oscillations = 9.0 J • angular frequency of oscillations = 1.2 rad s^{-1} **or** period of oscillations = 5.1 s **or** frequency of oscillation = 0.19 Hz • maximum speed of block = 0.73 m s^{-1} • mass of block = 33 kg	**B3**
5(c)	U-shaped curve resting on h axis (with minimum at E_P = 0)	**B1**
	curve from h = 0.8 m to h = 2.0 m, with minimum E_P shown at h = 1.4 m	**B1**
	both end-points of curve shown at E_P = 9.0 J	**B1**

Question	Answer	Marks
6(a)(i)	conversion from a.c. to d.c.	**B1**
6(a)(ii)	smoothing	**B1**
6(b)(i)	$A = 12\text{ V}$	**A1**
	$B = 2\pi / (20 \times 10^{-3})$ $= 310\text{ rad s}^{-1}$	**A1**
6(b)(ii)	full-wave (rectification)	**B1**
6(b)(iii)	four diodes shown, with correct circuit symbols	**B1**
	four diodes correctly connected to form a bridge rectifier	**B1**
6(b)(iv)	$V = V_0 \exp(-t / \tau)$ **or** $V = V_0 \exp(-t / RC)$ **and** $\tau = RC$	**C1**
	$8.0 = 12 \exp(-7.3 \times 10^{-3} / \tau)$	**C1**
	$\tau = 0.018\text{ s}$	**A1**
6(c)	time constant = RC	**C1**
	$R = (0.018 / 570 \times 10^{-6})$ $= 32\ \Omega$	**A1**

Question	Answer	Marks
7(a)	• force per unit length • force per unit current • length / current perpendicular to field *1 mark for any two points, 2 marks for all three points.*	**B2**
7(b)(i)	$F = BQv$	**B1**
7(b)(ii)	arrow at Y pointing vertically upwards	**B1**
7(b)(iii)	upwards deflection showing circular path	**B1**
7(c)(i)	electric field applied vertically downwards (may be shown on a labelled diagram)	**B1**
	electric force on particle in opposite direction to magnetic force (may be shown on a labelled diagram)	**B1**
	particle undeflected when magnitudes of electric and magnetic forces are equal	**B1**
7(c)(ii)	$EQ = BQv$	**B1**
	$v = E / B$	**A1**

Question	Answer	Marks
8(a)	wavelength associated with a moving particle	**B1**
8(b)	$\lambda = h / p$	**C1**
	$= (6.63 \times 10^{-34}) / (9.11 \times 10^{-31} \times 4.9 \times 10^{7})$ $= 1.5 \times 10^{-11}$ m	**A1**
8(c)	similarity: *any one point from*: • same mass • same <u>magnitude</u> of charge • both leptons	**B1**
	difference: *any one point from*: • electron has negative charge, positron has positive charge • positron is anti-particle of electron • electron is a particle, positron is an anti-particle	**B1**
8(d)(i)	(pair) annihilation	**B1**
8(d)(ii)	their mass gets converted into energy	**B1**
	(their mass–energy) becomes the energy of the gamma photons	**B1**
8(d)(iii)	they travel in opposite directions to conserve momentum	**B1**
8(d)(iv)	kinetic energy $= \frac{1}{2} \times 9.11 \times 10^{-31} \times (4.9 \times 10^{7})^2 = 1.1 \times 10^{-15}$ J	**A1**

Question	Answer	Marks
8(d)(v)	$E = mc^2$	**C1**
	$E = hc / \lambda$ **or** $E = hf$ **and** $c = f\lambda$	**C1**
	$(1.1 \times 10^{-15}) + (9.11 \times 10^{-31} \times (3.00 \times 10^8)^2) = (6.63 \times 10^{-34} \times 3.00 \times 10^8) / \lambda$ $\lambda = 2.39 \times 10^{-12}$ m	**A1**

Question	Answer	Marks
9(a)	number of nuclear disintegrations per unit time	**B1**
9(b)	activity is proportional to the number of undecayed nuclei	**B1**
	activity = (–) rate of change of number of undecayed nuclei	**B1**
	N is proportional to the rate of change of *N* (so exponential variation)	**B1**
9(c)(i)	$120 = 180 \exp(-\lambda \times 8.4)$	**C1**
	$\lambda = 0.048\ \text{min}^{-1}$	**A1**
9(c)(ii)	half-life = ln 2 / 0.048 = 14 min	**A1**
9(c)(iii)	line with negative gradient throughout, starting at (0, 180)	**B1**
	curve with negative gradient passing through (8.4, 120)	**B1**
	curve with decreasing negative gradient, from $t = 0$ to $t = 24$ min, passing through (14, 90)	**B1**

Question	Answer	Marks
10(a)	speed is (directly) proportional to distance	**M1**
	speed is speed of recession of galaxy from an observer, and distance is the distance of the galaxy from the observer	**A1**
10(b)(i)	galaxy is receding from the Earth	**B1**
	observed wavelength is redshifted from emitted wavelength	**B1**
10(b)(ii)	$\Delta\lambda / \lambda = v / c$ $(4.91 - 4.62) / 4.62 = v / (3.00 \times 10^8)$	**C1**
	$v = 1.9 \times 10^7\ \text{m s}^{-1}$	**A1**
10(b)(iii)	wavelength (of maximum intensity) is inversely proportional to temperature	**B1**
	observed wavelength too high, so determined temperature too low	**B1**
10(c)	$v = H_0 d$	**C1**
	$d = (1.9 \times 10^7) / (2.3 \times 10^{-18})$ $= 8.3 \times 10^{24}$ m	**A1**